

Influence of sintering parameters on the microstructure and mechanical properties of nanosized 3Y-TZP ceramics

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Abstract

The fracture toughness of 3Y-TZP ceramics obtained from a nanocrystalline powder with an optimized microstructure and highly transformable tetragonal grains was investigated. Samples of ZrO₂-3 mol% Y₂O₃ were sintered at temperatures between 1250 and 1400 °C, with isothermal holding times of up to 16 h. Samples sintered at 1250 °C exhibited relative densities ranging between 92% and 98%, which increased with increasing isothermal duration, while samples sintered at 1300, 1350, or 1400 °C achieved densification higher than 98% for all isothermal treatments. Crystallographic analysis indicated the presence of a highly transformable ZrO₂-tetragonal phase ($c/a\sqrt{2} = 1.0148-1.0154$) for all conditions studied. The average grain size ranged from 0.18±0.04 μm (1250 °C-0 h) to 0.64±0.08 μm (1400 °C-16 h), indicating activation energy of 141.3 kJ/mol for grain growth and a growth exponent of 2.8. Both Vickers hardness (1025 to 1300 HV) and fracture toughness (4.0 to 7.8 MPa.m^{1/2}) increased with increasing sintering temperature and time due to increased densification, reduced porosity, and maintenance of potentially high fracture toughness by the t→m phase transformation. **Keywords:** 3Y-TZP, nanosized powder, sintering, microstructure, t→m transformation toughening mechanism.

INTRODUCTION

The use of yttria-stabilized tetragonal zirconia-based ceramics, ZrO₂ (Y₂O₃), also called Y-TZP, is widespread in the field of dentistry, among others, due to its biocompatibility, aesthetic characteristics and, in particular, its excellent mechanical properties [1, 2]. In particular, a high fracture toughness results from the tetragonal to monoclinic (t→m) phase transformation that occurs in the wake field of a propagating crack and which is accompanied by a volumetric expansion of around 4% to 5% [3, 4]. This phase transformation in turn, generates stress fields in the ceramic matrix, which hinders crack propagation, thus improving fracture toughness. In general, ceramics with an average grain size of less than 100 nm are called nanocrystalline ceramics. With reduced grain sizes in the sintered body, nanocrystalline ceramics may have different properties and considerable merits compared to conventional ceramics [5]. In some cases, nanocrystalline ceramics are used as a base material for the development of components and sometimes are applied as a secondary phase (reinforcement) added to ceramic matrices, with the intention to improve the fracture toughness and sinterability of these materials. However, there are some problems associated with the use of nanocrystalline ceramic powders, in particular the difficulty of eliminating aggregates and agglomerates, as well as the difficulty of compaction and control of grain growth during the sintering process. Starting in the 1990s, a number of companies developed easily compressible ZrO₂ nanosized powders, stabilized with 3 mol% Y₂O₃, with added binders [6, 7]. Due to this scenario, a growing research interest exists to control grain growth during sintering and to evaluate the effects on the mechanical properties.

Complete densification of 3Y-TZP ceramics using micrometric or sub-micrometric powders can be achieved at sintering temperatures of 1500 °C or higher and isothermal holding periods. As a result, the increased diffusion may cause grain growth of the ZrO₂ grains and lead to a loss of stability of the tetragonal phase, even allowing for heterogeneous grain growth. In consequence, a population of low tetragonality t'-ZrO₂ grains is generated [8, 9] with a concomitant loss of transformation ability (fracture toughness) when the material is subject to tensile stresses at the crack tip of a propagating crack. The use of nanosized ceramic powders is of great scientific and technological interest due to their increased sinterability, which may reduce the sintering temperature and/or time, also resulting in extremely fine-grained microstructures with improved mechanical properties. In the case of ZrO₂-based ceramics, using nanosized particles, an improvement of the fracture toughness by microcracking has been reported [6-9]. Therefore, the sintering cycle has to be carefully chosen in order to take advantage of the unique properties of nanostructured materials. Previous studies [9, 10] indicate that the effectiveness of the t→m phase transformation of Y-TZP ceramics is associated with the transformability of the tetragonal grains present in the sintered material. This transformability is associated with shearing stresses present in the tetragonal anisotropic phase, which is more pronounced for structures with a high ratio of the lattice parameters 'c' and 'a'. Thus, maximizing the fraction of transformable tetragonal grains in the region ahead of the crack tip allows for increased fracture toughness, and specifically for 3Y-TZP ceramics, grain sizes between 0.1 and 1.0 μm should be targeted [9], as grain sizes outside this range notably present low transformability or spontaneous transformation [9].

In this work, the aim was to specifically investigate the effects of the sintering parameters on a 3Y-TZP

nanoparticulate powder that allows for the unique metastability of highly transformable ZrO₂-tetragonal grains, developing microstructures with an average grain size smaller than 1 μm and correlating the densification, amount of tetragonal ZrO₂ phase, and microstructure (grain growth) with the resulting mechanical properties of low-temperature sintered 3Y-TZP ceramics.

EXPERIMENTAL PROCEDURE

The starting powder used was a commercial nanosized 3Y-TZP powder (TZ-3YE, Tosoh) with a specific surface of 16.2±2.0 m²/g, an average crystallite size of 40 nm, and containing 3.6 wt% of binder.

Dilatometry and sintering of Y-TZP samples: 3Y-TZP bars (4x4x8 mm) and discs (Ø12x3 mm) were compacted by cold uniaxial pressing under applied pressures between 12 and 74 MPa for 60 s. The samples were sintered under two different conditions: a) the sinterability of the Y-TZP specimens was evaluated by dilatometry using a dilatometer (DIL-402C, Netzsch) under argon flux throughout the heating and cooling cycle, adopting a heating rate of 2 °C/min up to the maximum sintering temperature of 1250, 1300, 1350, or 1400 °C; the shrinkage was measured by a linear variable differential transducer (LVDT) with a sensitivity of 0.01 mm; and b) solid-state sintering, using an electrically heated furnace (F1650, Maitec), with a heating rate of 2 °C/min up to the maximum sintering temperature of 1250, 1300, 1350, or 1400 °C with various isothermal holding times up to 16 h. Cooling to room temperature was done at a rate of 10 °C/min.

Density and phase analysis: the relative green density of the compacted samples was determined by the ratio of the geometrical density of the samples and the theoretical density of the material (ρ_{th}) of 6.05 g/cm³. The density of the sintered samples was determined by the immersion method, using Archimedes' principle. The residual porosity was calculated by the following equation [11]:

$$P(\%) = [1 - (\rho/\rho_{th})] \cdot 100 \quad (A)$$

The phase composition of the starting powders and sintered samples was determined by X-ray diffraction analysis using a diffractometer (XRD-6000, Shimadzu). The analysis was conducted with CuKα radiation in the 2θ range of 20° to 80° with a step width of 0.05° and a counting time of 3 s/position. The X-ray diffraction peaks were identified by comparison with the JCPDS files [12]. The monoclinic zirconia content at the sample surface (F_M) was calculated by the integrated peak areas of the planes (1̄11)_M and (111)_M planes of the monoclinic phase, in relation to the peak area of the (101)_T plane of the tetragonal phase, according to [13]:

$$X_M = \frac{(\bar{1}11)_M + (111)_M}{(\bar{1}11)_M + (111)_M + (101)_T} \quad (B)$$

$$F_M = \frac{1.311 \cdot X_M}{1 + 0.1311 \cdot X_M} \quad (C)$$

The calculations of the lattice parameters were done with the Rietveld refinement technique, using the FullProf Suite 3.0 software [14-16]. The tetragonality, expressed by the c/a ratio of the ZrO₂ grains, of samples sintered at 1250 °C/0 h and 1400 °C/16 h was determined by the model proposed by Krogstad et al. [17, 18] valid for ZrO₂ stabilized with different amounts of Y₂O₃. After sintering, both surfaces of the specimens were ground and polished successively with 9, 6, and 1 μm diamond suspensions. The microstructures of the sintered samples were observed using a scanning electron microscope (SEM, JSM-5310, Jeol). Samples were thermally etched at 1250 °C for 15 min with a heating rate of 25 °C/min. The grain size distributions of the sintered samples were determined using the ImageJ software. A population of at least 400 grains was analyzed for each sintering condition studied.

Mechanical properties: Vickers hardness of the sintered 3Y-TZP specimens was measured using a hardness testing equipment (Time Hardness). The polished surface of the samples was indented with a load of 1000 gf for 30 s, conducting 30 measurements (n=30) for each sintering condition studied. Furthermore, the fracture toughness (K_{Ic}) was also measured by the crack length emanating from the Vickers indentation marks, as proposed by Niihara et al. [19]:

$$K_{Ic} = 0.0089 \left(\frac{E}{HV} \right)^{2/5} \left(\frac{F}{a \cdot l^{1/2}} \right) \quad (D)$$

where K_{Ic} is the fracture toughness (MPa·m^{1/2}), l is the length of the crack, measured from the tip of the indentation until the tip of the crack (μm), a is the half-length of the indentation diagonal (μm), HV is the Vickers hardness (GPa), E is Young's modulus of 3Y-TZP (195 GPa), and F is the indentation load used in the Vickers hardness test (N).

RESULTS AND DISCUSSION

Characterization of starting-powder: Fig. 1 shows the X-ray diffraction (XRD) pattern, the compaction curve, and an SEM micrograph of the zirconia powder used. A majority of the tetragonal (t-ZrO₂) phase was found in the starting powder, as well as 18 vol% of the monoclinic (m-ZrO₂) phase (Fig. 1a). It can be noted from Fig. 1b that the material reached its maximum green density level for compaction pressures above 60 MPa, remaining constant for higher pressures. This behavior is believed to be the result of the binder present in the commercial powder. Furthermore, the

SEM micrograph (Fig. 1c) shows that the starting powder consisted of spherical agglomerates with sizes ranging between 20 and 50 μm , which were possibly formed during a spray-drying process of the powders with 3.6 wt% of binder added in order to facilitate the compaction process of this powder.

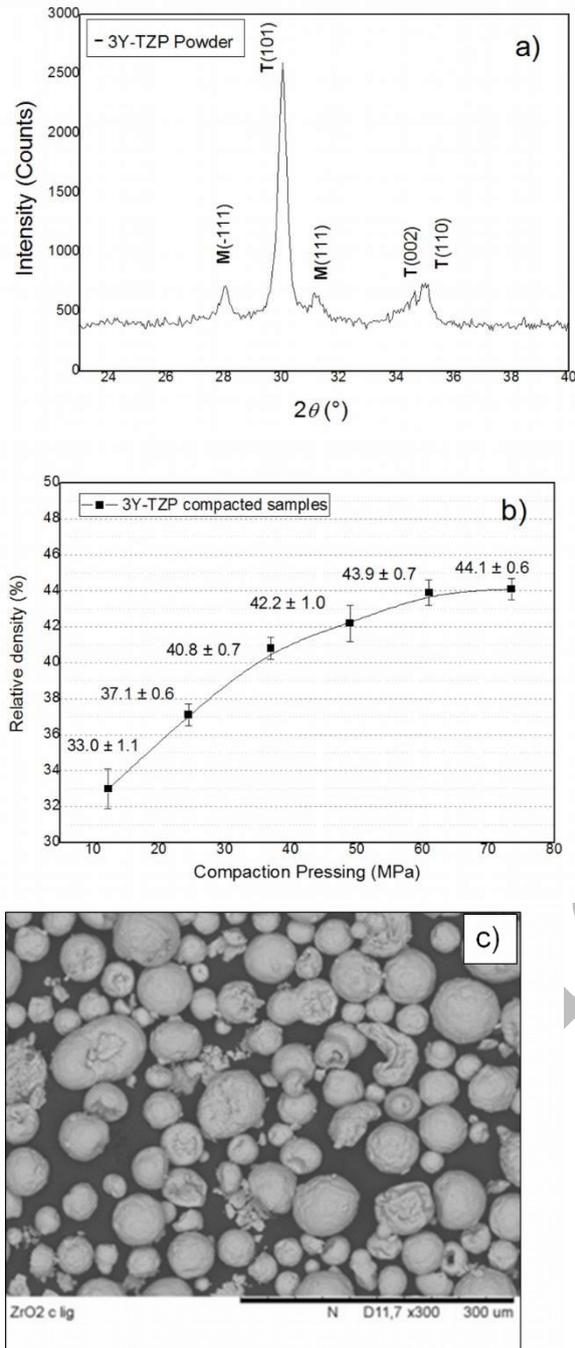


Figure 1: XRD pattern of the starting Y-TZP powder (a), green density as a function of the compaction pressure (b), and SEM micrograph of the Y-TZP powder highlighting agglomerates of varying size (c).

Dilatometry: Fig. 2a presents the linear shrinkage and shrinkage rate of a ZrO_2 sample as a function of the sintering temperature up to 1400 $^\circ\text{C}$, while Fig. 2b presents a comparison of the shrinkage rate of two samples compacted at the lowest and highest pressure of 12.3 and 73.5 MPa investigated in this work and sintered at 1400 $^\circ\text{C}$ with an isothermal holding time of 60 min. It was observed that at temperatures between 200 and 400 $^\circ\text{C}$, region I, a first shrinkage step occurred, corresponding to the elimination of the organic binder in the starting material. The highlighted region II comprised the effective onset of the densification process, where the solid-state sintering mechanisms acted eliminating porosity. In microparticulate Y-TZP ceramics, this region where densification usually begins is close to 1150 $^\circ\text{C}$ [20], while in nanocrystalline materials with a particle size of about 40 nm, neck formation, and consequent densification and shrinkage occurred at temperatures close to 980 $^\circ\text{C}$. It can be noted that before the sintering isotherm begins, the effective shrinkage gain with increasing compaction pressure was about 2-3%. Thus, for further studies in this work, a compaction pressure of 73.5 MPa for the preparation of samples was chosen.

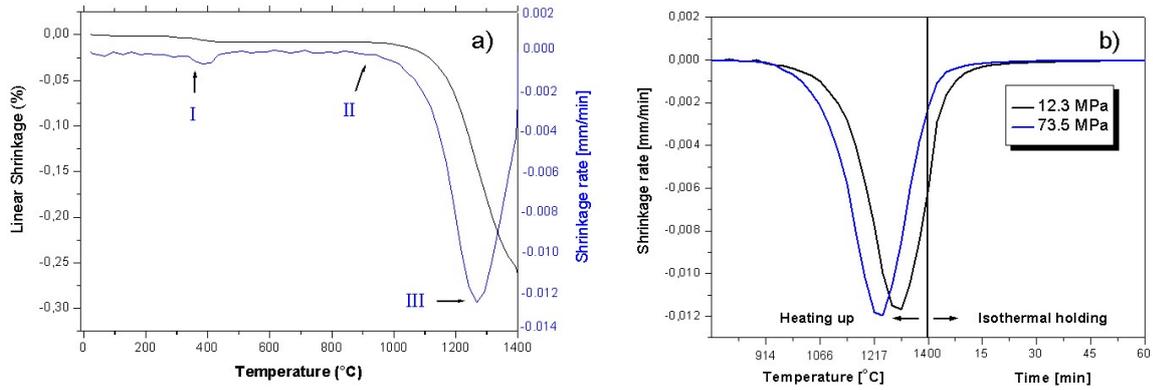


Figure 2: Linear shrinkage (dL/L_0) and shrinkage rate as a function of the temperature of typical compacted 3Y-TZP sample (a), and shrinkage rate of 3Y-TZP samples compacted with different pressures (final temperature of 1400 °C) (b).

Characterization of sintered bodies: the relative density and residual porosity results of the Y-TZP samples as a function of the sintering temperature are shown in Fig. 3. The extremely fine, nanometric particle size of the zirconia starting powder used allowed for high densification to be reached during sintering. Under all sintering temperatures studied, relative densities higher than 92% were achieved, and for the sintering temperatures of 1350 and 1400 °C, all samples exhibited relative density higher than 98%, even without isothermal holding (Fig. 3a). At lower temperatures, 1250 and 1300 °C, relative densities exceeding 98% were only obtained after an isothermal treatment for 16 or 8 h, respectively. Similarly, the residual porosity (Fig. 3b) was progressively reduced with increasing temperature and isothermal holding time.

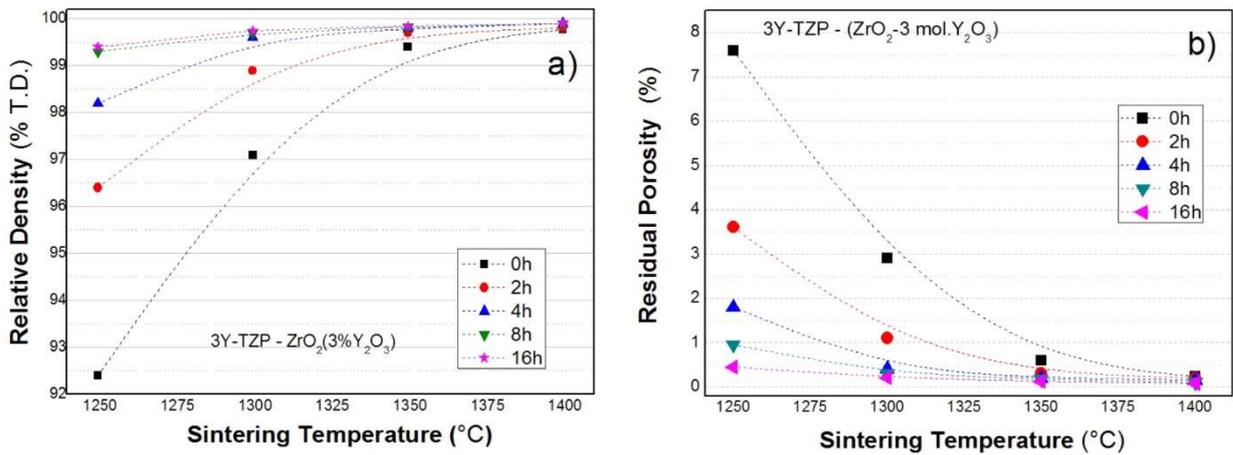


Figure 3: Relative density (a) and porosity (b) of 3Y-TZP sintered samples as a function of sintering temperature and isothermal holding time.

The X-ray diffraction patterns of samples sintered at different temperatures and isothermal holding times are shown in Fig. 4. Similar diffraction patterns are observed in Fig. 4 for all samples showing only diffraction peaks of the tetragonal ZrO₂ phase, independently of the sintering temperature or holding time investigated. Furthermore, no monoclinic ZrO₂ was detected after sintering, indicating the complete conversion of this phase present in the starting powder (Fig. 1a) into tetragonal ZrO₂. The results of the Rietveld refinement for the t-ZrO₂ lattice parameters and the calculated tetragonality of samples sintered at 1250 °C/0 h, 1350 °C/8 h, and 1400 °C/16 h are summarized in Table I. Scott [21], reporting on the ZrO₂-Y₂O₃ binary system, indicates that ZrO₂ ceramics stabilized with 3 mol% Y₂O₃ present the tetragonal phase with some associated cubic phase. However, in recent studies [17, 18], mathematical models associated with the X-ray diffraction technique propose that an intermediate phase called t'-ZrO₂, composed of grains containing network parameters 'c' approaching 'a', can coexist with the stabilized t-ZrO₂ phase when 3Y-TZP is sintered at temperatures and times that allow for the migration of a part of Y₂O₃ to specific grains t', reported as 'semi-cubic'. In this way, a modification of the original unit cell structure may occur, which changes the relationship between the parameters c and a of certain grains. Thus, after determining the crystallographic parameters of sintered materials at different temperatures and holding times, the results shown in Table I indicated that the tetragonality (c/a ratio) of the materials underwent little variation, with values of c/a√2 of 1.015, regardless of the sintering conditions. As a result, the tetragonality of the grains was not significantly altered by the sintering temperature and holding time studied.

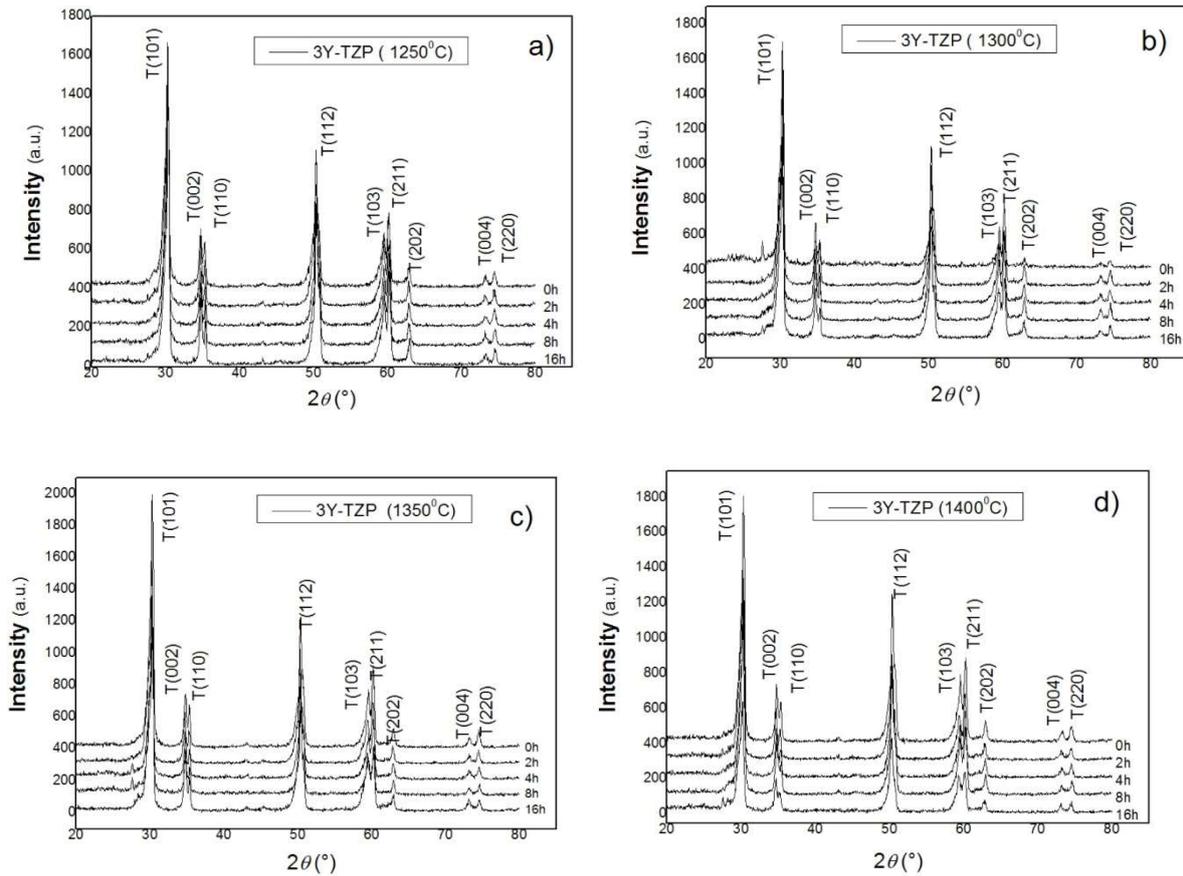


Figure 4: XRD patterns of 3Y-TZP samples sintered at 1250 °C (a), 1300 °C (b), 1350 °C (c), and 1400 °C (d), with isothermal holding times of 0, 2, 4, 8, and 16 h.

Table I - Rietveld refinement of 3Y-TZP crystal structure after sintering at different temperatures.

Parameter	1250 °C - 0 h	1350 °C - 8 h	1400 °C - 16 h
Lattice parameters: tetragonal	$a = 3.5956 \text{ \AA}$	$a = 3.5913 \text{ \AA}$	$a = 3.5904 \text{ \AA}$
ZrO ₂ (t-ZrO ₂); space group	$c = 5.1601 \text{ \AA}$	$c = 5.1570 \text{ \AA}$	$c = 5.1548 \text{ \AA}$
$P4_2/nmc$ (137)	$V = 66.711 \text{ \AA}^3$	$V = 66.513 \text{ \AA}^3$	$V = 66.451 \text{ \AA}^3$
Tetragonality ($c/a\sqrt{2}$) [17, 18]	1.0148	1.0154	1.0152
χ^2	2,58	4.08	5.72

Fig. 5 shows representative micrographs of samples sintered at 1250 and 1400 °C with isothermal holding times of 0, 2, 4, 8, and 16 h. Furthermore, Fig. 6a presents the average grain size as a function of the sintering temperature and isothermal holding time, and Fig. 6b shows a correlation between the grain size and the relative density of the sintered samples. The increase of the average grain size of the zirconia grains was observed with the increase of the sintering temperature and isothermal holding time (Fig. 6a). The grain size varied between 0.18 ± 0.04 and $0.45 \pm 0.07 \mu\text{m}$ for samples sintered at 1250 °C without and 16 h isothermal holding time, respectively, and between 0.22 ± 0.06 and $0.64 \pm 0.08 \mu\text{m}$ for samples sintered at 1400 °C without and 16 h isothermal holding time. The calculated grain growth exponent (n) and the activation energy for grain growth (Q_{gg}) were $n=2.8$ and $Q_{gg}=141.3 \text{ kJ/mol}$ for the Y-TZP powder studied, indicating that grain growth was controlled by grain boundary diffusion [7, 20, 22]. In Fig. 6b, the average grain size is correlated to the relative density. The graph indicates that grain growth occurred simultaneously with the elimination of residual porosity at all sintering temperatures studied. Analyzing the shrinkage behavior obtained by dilatometry (Fig. 2), it was observed that at about 980 °C, the densification process began, due to the high reactivity of the nanocrystalline powder, with crystallite sizes of around 40 nm. From this temperature, until the final sintering temperatures (1250 to 1400 °C) were reached, diffusional mechanisms acted in the material, reducing the free internal surface area and eliminating porosity, besides initiating grain growth. On the other hand, the materials reached high densification levels, with relative densities higher than 92% and presenting average submicron grain sizes between 0.18 and 0.22 μm , with low size variations between sample groups.

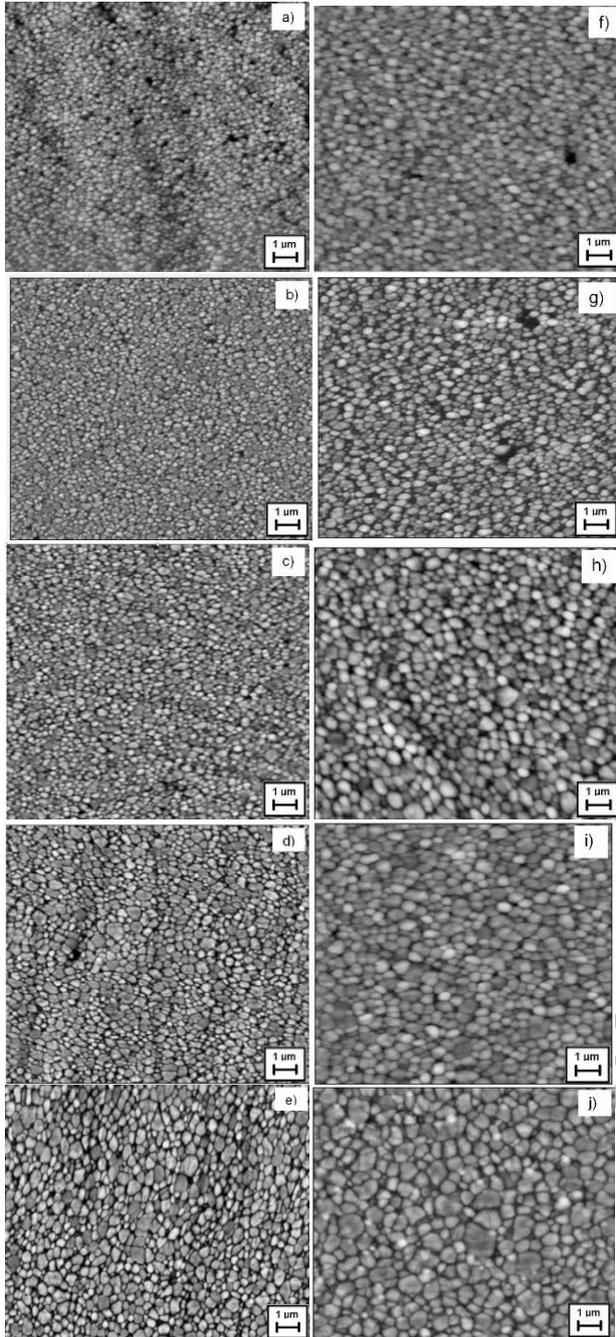


Figure 5: SEM micrographs of Y-TZP samples sintered at: 1250 °C for 0 h (a), 2 h (b), 4 h (c), 8 h (d), and 16 h (e); and 1400 °C for 0 h (f), 2 h (g), 4 h (h), 8 h (i), and 16 h (j).

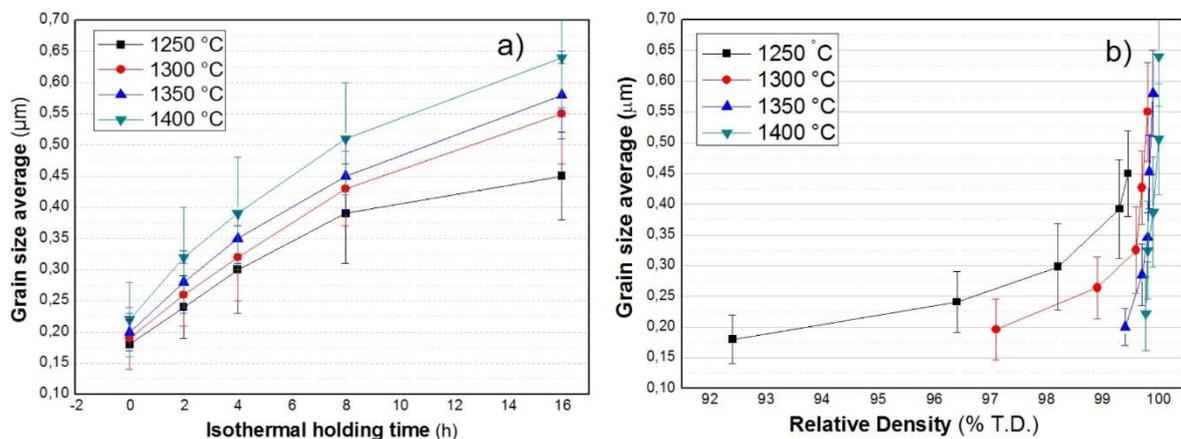


Figure 6: Grain size as a function of isothermal holding time at different temperatures (a), and the correlation between grain size and relative density (b) for Y-TZP ceramics.

In general, when the relative density exceeds 90% [23-28] during the sintering of submicron-sized particles, the bonding between the particles due to grain boundary formation is well developed, and most pores in the sintered body are closed. Furthermore, grain growth occurs until reaching the final sintering temperature and also during the isothermal holding time, as was the case in this work. Densification also continues to increase, although the few and well dispersed residual pores may still hinder grain growth kinetics during this stage. In consequence, first, the final average grain sizes of the sintered samples were very close and with only small variations due to the very small ZrO₂ grains formed in the early sintering stage of nanocrystalline Y-TZP powders; second, the residual porosity (Fig. 3b) was low at the moment the isothermal threshold began (t=0); and, third, the sintering temperatures adopted in this study (1250-1400 °C) allowed only a moderate diffusivity compared to the usual sintering temperatures of 1500 to 1600 °C [6, 20]. Therefore, the sintering of these nanoparticulate powders results in normal grain growth during the isothermal holding time, and refined microstructures were obtained under all sintering conditions studied.

Mechanical properties: in Fig. 7a, the Vickers hardness, and in Fig. 7b, the fracture toughness of sintered samples are presented as a function of the sintering parameters adopted. Furthermore, the relationship between the number of grains per unit area (grain density) as a function of temperature and isothermal holding time is shown in Fig. 7c. A slight increase in hardness was observed with increasing duration of the isotherm for all sintering temperatures (Fig. 7a). This behavior was associated with the increase in densification and reduction in porosity, as illustrated in Fig. 3. Porosity exponentially reduces the hardness of the material, and therefore, its elimination improves the final hardness of the sintered body. Finally, maximum hardness ranged from 1250 to 1300 HV, which are typical values for Y-TZP-based ceramics [29, 30]. The fracture toughness (Fig. 7b) also increased with increasing sintering temperature and isothermal holding time. The fracture toughness of zirconia based-ceramics is associated with three material characteristics: i) tetragonal phase content, ii) ZrO₂ grain size, and iii) residual porosity. Microstructural analysis revealed average grain sizes ranging from 0.18 to 0.22 μm for sintered materials without an isothermal threshold and 0.45 to 0.64 μm for prolonged isothermal treatments. Different authors claim, based on previous studies, that dense, mostly tetragonal 3Y-TZP ceramics may have grains with different degrees of toughening effect by the t→m phase transformation [31-34]. Specifically, when the grain size is less than or near 100 nm, the grains are highly metastable, tending to remain tetragonal even when exposed to a stress field generated by the crack propagation, thus without undergoing the t→m transformation toughening mechanism. On the other hand, materials with zirconia grains larger than 1 μm tend not to exhibit thermodynamic metastability, leading to a spontaneous phase transformation or, in some cases, to a further transformation into the cubic phase, with depletion of Y in neighboring grains and generation of monoclinic ZrO₂. Consequently, materials exhibiting an average grain size close to these two limits tend to have reduced mechanical properties, especially with regard to fracture toughness. Fig. 7c shows the grain density as a function of sintering temperature and time. It was noted that without isotherm or for short isothermal periods, the number of grains per area was higher, regardless of the temperature. However, as already shown, many of these grains were of a very small size. For isothermal holding times longer than 4 h, the grain density tended to stabilize, with grain sizes ranging from 0.35 to 0.64 μm. The third factor that significantly influences the fracture toughness of Y-TZP ceramics is the residual porosity (Fig. 3b). In the case of our work, an inverse proportional behavior of fracture toughness with the reduction of porosity was found, as confirmed by other works [35-37].

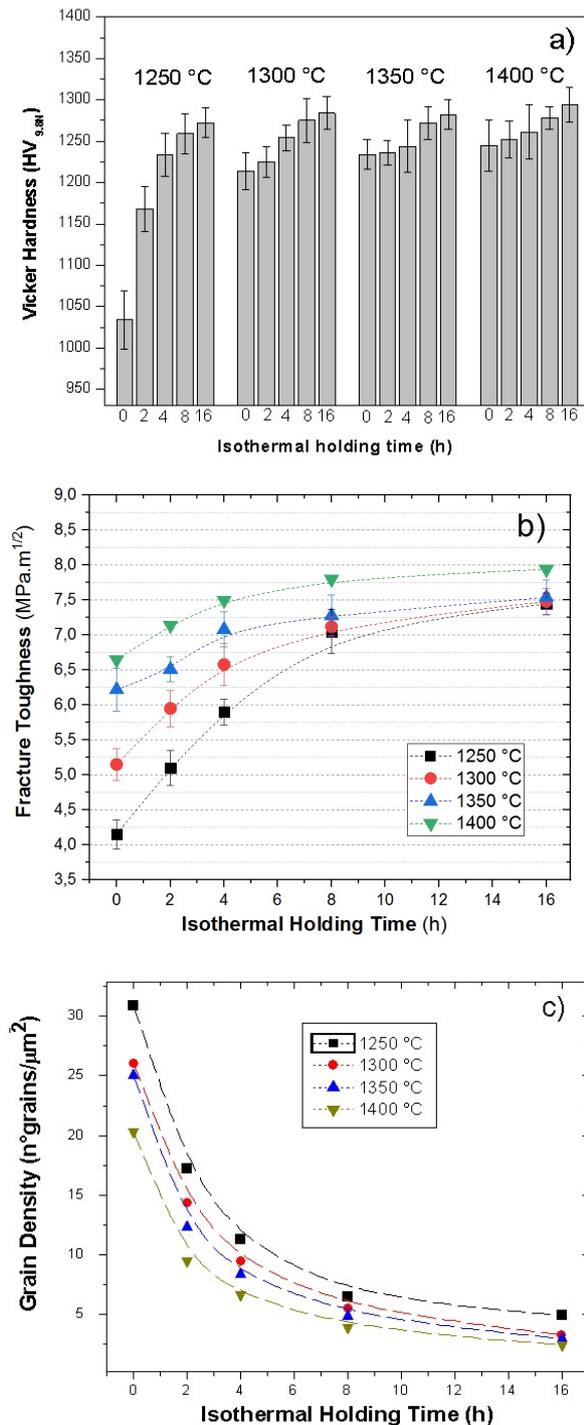


Figure 7: Vickers hardness (a), fracture toughness (b), and grain density (c) as a function of sintering temperature and isothermal holding time for the 3Y-TZP sintered samples.

It is well established that the fracture toughness of tetragonal zirconia (Y-TZP) based ceramics is the result of its peculiar $t \rightarrow m$ phase transformation mechanism [9, 38]. This transformation occurs basically in two stages: transition of the tetragonal into the monoclinic structure due to the displacement of the Zr^{4+} ions, and the diffusion of oxygen ions to the oxygen sites in the monoclinic structure, causing monoclinic, longitudinal sheets grow within the tetragonal grain, just as they grow laterally due to the lateral migration of O^{2-} ions [39]. As a crack propagates, tensile stress is associated with the crack opening. The tetragonal grains adjacent to the crack tip undergo compression and create a stress zone associated with the applied mechanical stress. The variation of the total free energy for the transformation to occur is the energy balance between the variation of the free chemical energy, variation of surface free energy, and density of interaction energy (associated with the application of external energy) [40, 41]. In the specific case of this work, for all sintered samples studied, the mechanical tests were carried out at room temperature, the chemical composition referring to the alloying oxide (3 mol% Y_2O_3) remained constant, and the energy of interaction was considered constant because the same indentation load was adopted in the tests. Thus, the transformation occurred mainly due to the factor related to the free surface energy; in other words, the tetragonal microstructure was the dominating factor in the transformation.

Considering that the t - ZrO_2 grains were the main parameter responsible for the fracture toughness of Y-TZP

ceramics, the extent of the transformation region (shielding) around the crack tip was directly linked to the volumetric fraction of transformable t-ZrO₂ grains. In the specific case of this work, we adopted a microstructural design that allowed for a complete formation of transformable tetragonal grains, confirmed by the X-ray diffraction analysis that indicated a stable tetragonality of the grains, regardless of the sintering conditions adopted. As the tetragonality was not influenced within the limits of this study, the shear stress and consequent deformation induced by twinning did not significantly affect toughness. The results indicated that porosity was progressively reduced in the sintered Y-TZP ceramics due to the increase in temperature and isothermal holding time. Furthermore, it is expected that the elastic modulus undergoes proportional reduction with increasing porosity and that porosity reduces the effective phase transformation zone around the crack tip (shielding) because they represent voids which retain part of the cascade effect of shearing or compressive stresses resulting from transformed grains located in this region. The increment of toughness by plane deformation (ΔK_C) can be expressed by [42, 43]:

$$\Delta K_C = 0.22 \cdot f \cdot E \cdot \varepsilon^t \cdot \frac{\sqrt{h}}{1-\nu} \quad (E)$$

where f is the volume fraction of the tetragonal phase in the transformation zone, E is Young's modulus, ε^t the dilatational deformation involved, h the size of the transformation zone, and ν the Poisson's ratio. According to this model, considering the microstructure and the porosity of the Y-TZP ceramics developed in this work, it can be stated that sintering at lower temperatures and shorter holding times results in higher porosity and smaller average grain size and therefore a reduction in the parameters f , E , and h (Eq. E). In addition, as there are no variations in the tetragonality and, therefore, the transformability of tetragonal grains can be considered similar, the parameter ε^t can be considered constant, as well as the Poisson's ratio. On the other hand, the increase in the size of the tetragonal grains, on this dimensional scale, leads to a small increase in the effective transformation zone around the crack tip. Thus, within the experimental limits of this article, the toughening and, consequently, the fracture toughness of these materials containing smaller grain size and greater porosity is theoretically lower than dense materials with larger zirconia grains, corroborating the results presented in Fig. 7b.

CONCLUSIONS

The fracture toughness of ceramics obtained from a nanocrystalline 3Y-TZP powder and sintering between 1250 and 1400 °C was investigated. Samples sintered at 1250 °C showed relative densities between 92% and 98%, depending on the isothermal holding time. For sintering at 1350 and 1400 °C, all samples exhibited relative density higher than 98%, independent of an additional isothermal holding. The analysis by Rietveld's refinement indicated that the only crystalline phase present under all sintering conditions studied was t-ZrO₂, whose tetragonality enabled high transformability during activation of the t→m transformation toughening mechanism. In addition, the microstructural analysis revealed that the average grain sizes ranged from 0.18 to 0.22 μm for sintered materials without an isothermal threshold and 0.45 to 0.64 μm for prolonged isothermal treatments of 16 h. The hardness increased proportionally with the reduction of the residual porosity, while the fracture toughness presented a progressive increase related to the increase in the sintering temperature as well as in the isotherm sintering (4 to 8 MPa.m^{1/2}). This behavior was due to the increase in the effective transformation zone around the crack tip, which is reported to increase with increasing average grain size, without losing its transformability, as well as the gradual reduction of residual porosity. Thus, an efficient strategy to densify nanoparticulate powders at low temperatures, obtaining a high fracture toughness, may be the use of isothermal holding times that result in refined and highly transformable microstructures.

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Segue em anexo imagens jpeg, exceto Figs 2 e 7 que estão melhores no Word

Página 2 tem símbolo destacado com fundo amarelo.

Figure 1a, eixo x: 2θ (degree)

Figure 1a, eixo y minúscula e singular: Intensity (count)

Figure 1b, eixo x: Compaction pressure (MPa)

Figure 2, eixo y, trocar vírgula por ponto para separador de decimal de número

Figure 4, eixo x: 2θ (degree)

Figure 5, se for deixar as imagens na vertical, como mostrado, deixar imagens a,b,c,d do lado esquerdo e e,f,g,h do lado direito (como no Word)

Figure 6, eixo y: Average grain size (μm)